The Role of Hydrocarbons in the lonospheres of the Outer Planets

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The role of hydrocarbons as a possible sink for H^+ and H_3^+ ions in the lower ionosphere of the outer planets is examined. Calculations indicate that H^+ and H_3^+ are efficiently converted to hydrocarbon ions on reaction with methane. The terminal ions, CH_5^+ and $C_2H_5^+$ are rapidly neutralized in dissociative recombination with electrons. Extreme ultraviolet photolysis of hydrocarbons as a potential additional source of lower elevation ions is investigated.

INTRODUCTION

McElroy (1973) first raised the possibility that protons in the Jovian ionosphere may be efficiently removed at lower altitudes by reaction with methane. In our last paper on the model ionospheres of Saturn, Uranus and Neptune (Atreya and Donahue, 1975) we mentioned that H^+ and H_3^+ would rapidly combine with CH_4 to form CH_5^+ and $C_2H_5^+$ if the reaction rates were comparable to the gas kinetic rates. The recently measured values of rate constants for hydrocarbon reactions with H^+ , H_3^+ , and H^+ (Huntress, 1975) confirm our suspicions. In this follow-up note we present the result of calculations based on a chemical model which takes into consideration such hydrocarbon reactions. Results are first presented for Saturn and potential ionization processes at lower elevations for both Saturn and Jupiter are discussed. As we mentioned earlier (Atreya and Donahue, 1975) the methane mixing ratio on Uranus and Neptune is not firmly established yet, therefore we exclude these planets from our present discussion. The model calculations, however, are general and can be easily extended to these other outer planets.

Models

We adapt a medium K (eddy diffusion coefficient, $K = 2 \times 10^{6} \text{ cm}^{2} \text{ sec}^{-1}$) neutral

Copyright \bigcirc 1975 by Academic Press, Inc. All rights of reproduction in any form reserved. Printed in Great Britain model atmosphere for Saturn as we did in our last paper (Atreya and Donahue, 1975). This is done for the sake of consistency and because the value of K is quite uncertain. The chemical model is presented in Table I. All reactions other than those with the hydrocarbons are taken from Atreva and Donahue (1975). Hinteregger (1970) solar euv fluxes were scaled to the planet, and divided by a factor of two to construct diurnal average models. A solar zenith angle of 60° was assumed and the wavelength interval between 0 and 960Å was divided into grids of 5Å size; appropriate crosssections were averaged in the grid interval. References for the cross-sections are found elsewhere (McElroy, 1973). Ion densities calculated under assumptions of photochemical equilibrium are shown in Fig. 1. One notes, as before (Atreya and Donahue, 1975), that H^+ is the dominant ion above 160km. In this region removal of H⁺ takes place principally by electron recombination. Below 160km, conversion of H^+ to H_3^+ by three-body association reaction (e10) begins to be significant. Below 140km, methane begins to play a major role in converting H_1^+ (and the remaining H^+) to the hydrocarbon ions. There is a sharp switchover in role as major and minor ions from H_3^+ and H^+ to $C_2H_5^+$ and CH_5^+ . $C_2H_5^+$ and CH_5^+ are rapidly neutralized on dissociative recombination with electrons (r6 and r7).

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TABLE I

IMPORTANT REACTIONS IN THE IONOSPHERES OF THE OUTER PLANETS

Reaction	ı		
number	Reaction	Rate constant ^a	Reference
n	Ion production :		
pl	$H_2 + h\nu \rightarrow H_2^+ + e$		
$\mathbf{p2}$	\rightarrow H ⁺ + H + e		
p3	$H_2 + e \rightarrow H_2^+ + 2e$		
p4	$\rightarrow H^{+} + H + 2e$		McElroy (1973)
p5	$H + h\nu \rightarrow H^+ + e$		5 ()
p6	$H + e \rightarrow H^+ + 2e$		
p7	$He + h\nu \rightarrow He^+ + e$		
p8	$\mathrm{He} + \mathrm{e} \rightarrow \mathrm{He}^{+} + 2\mathrm{e}$		
	Ion exchange:		
el	$H_a^+ + H_a \rightarrow H_a^+ + H_b$	2.0×10^{-9}	
e2	$H_2^+ + H \rightarrow H_2 + H^+$	$\sim 1.0 \times 10^{-10}$	
e3	$H_2 + H_2 \rightarrow H_2 + H_1^+$	~ 20%	Atreva and Donahue
e4	\rightarrow HeH ⁺ + H (sum	10×10^{-13}	(1975)
e5	\rightarrow He + H + H+	> 80%	(1070)
-66	$H_0^+ \pm CH_1 \rightarrow CH^+ \pm H_2 \pm H \pm H_0$	2.4×10^{-10}	
e7	$\rightarrow CH_1 + H_2 + H_1$	0.3×10^{-10}	
68	$\rightarrow CH_2 + H_2 + H_0$	6.3×10^{-11}	Huntross (1075)
_0	$\rightarrow OH_3 + H + H_0$	3.9×10^{-11}	110101055 (1878)
e10	$\rightarrow 011_4 + 116$ H ⁺ \pm H ₂ \pm H ₂ \pm H ₂	3.0×10^{-29}	Atreva and Donahuo
010		0.2 × 10	(1975)
ell	$H^+ + CH_4 \rightarrow CH_2^+ + H_2$	2.3×10^{-9}	$H_{\text{untress}}(1975)$
e12	$\rightarrow CH_1^+ + H$	1.5×10^{-9}	1101005 (1070)
e13	$HeH^+ + H_2 \rightarrow H_2^+ + He$	1.85×10^{-9}	Atreva and Donahue
010			(1975)
e14	$H_3^+ + CH_4 \rightarrow CH_5^+ + H_2$	2.4×10^{-9}	()
e15	$CH^+ + H_2 \rightarrow CH_2^+ + H_2^-$	1.0×10^{-9}	
e16	$CH_{2}^{+} + H_{2}^{-} \rightarrow CH_{3}^{+} + H_{2}^{-}$	7.2×10^{-10}	
e17	$CH_3^+ + CH_4^- \rightarrow C_2H_3^+ + H_2$	$9.5 imes 10^{-10}$	Huntress (1975)
e18	$CH_4^+ + CH_4 \rightarrow CH_5^+ + CH_3$	1.15×10^{-9}	
e19	$\mathrm{CH}_4^+ + \mathrm{H}_2 \rightarrow \mathrm{CH}_5^+ + \mathrm{H}$	4.1×10^{-11}	
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	ion removal/electron-ion recombination:		
rl	$H_3^+ + e \rightarrow H_3 + H_3$	3.8×10^{-7}	
$\mathbf{r2}$	$H_2^+ + e \rightarrow H + H$	$< 1.0 \times 10^{-8}$	
r3	$\mathrm{He}\mathrm{H}^{+}+\mathrm{e} \rightarrow \mathrm{He}+\mathrm{H}$	$\sim 1.0 \times 10^{-8}$	Atreya and Donahue (1975)
$\mathbf{r4}$	$H^+ + e \rightarrow H + h\nu$	6.6×10^{-12}	·/
$\mathbf{r5}$	$He^+ + e \rightarrow He + h\nu$	6.6×10^{-12}	
r6	$CH_5^+ + \Theta \rightarrow)$	1.9×10^{-6}	Rebbert et al. (1973)
$\mathbf{r7}$	$C_2H_5^+ + e \rightarrow$ neutral products	1.9×10^{-6}	Rebbert <i>et al.</i> (1973)

^a The rate constants are in units of $cm^3 sec^{-1}$ for two-body reactions, and $cm^6 sec^{-1}$ for three-body reactions.



FIG. 1. A model for Saturn's ionosphere with eddy diffusion coefficient, $K = 2 \times 10^6 \text{ cm}^2 \text{sec}^{-1}$. The vertical scale gives the height above a reference level where the hydrogen density is 10^{16} molecules cm⁻³.

DISCUSSION

Calculations indicate that CH₄ provides an important sink for H^+ and H_3^+ ions up to an altitude just above the methane turbopause. The important ions in this region are $C_2H_5^+$ and CH_5^+ . Hydrocarbons other than methane are not found to be important sinks of H^+ and H_3^+ . Strobel (1975) pointed out the potential importance of euv photolysis of CH_4 which results in CH_4^+ and CH₃⁺ ions below 945Å. Our calculations reveal that the absorption by H_2 in this region of the spectrum is so large that methane photoionization contributes no more than 3% to the electron density. Strobel (personal communication, 1974) also estimates an upper limit of 2×10^{-17} cm^2 for the value of photoionization cross-section of CH₃, whose measured ionization potential is 9.82 ± 0.04 eV (Elder et al., 1962). Solar Ly α is expected to be the principal ionizing flux for this methyl radical. Ionization from radiation other than Ly α should be insignificant. For Saturn, crude hydrocarbon models with $K = 2 \times 10^6 \,\mathrm{cm^2 \, sec^{-1}}$ give a value of CH₃ much too low for it to cause any visible change in the electron density profile due to CH_3^+ ion production. However, for purposes of illustration, we show, in Fig. 2, Jupiter's electron density profiles assuming Strobel's (1975) $K = 10^5$ (low K) and $K \propto 1/(M)^{1/2}$ (high K; $K = 3 \times 10^7$ at the turbopause) hydrocarbon models. The discussion of location and magnitude of the electron density maxima in the low and high K models is found elsewhere (Atreya et al., 1974). Here we simply wish to illustrate the influence of CH₃⁺ production on the ionospheric profile based on the chemical model listed in Table I. It can be seen that the CH₃⁺ production peaks at an altitude where CH₃ maxima occur and where optical depth in methane is about 0.3. CH_3^+ production drops off sharply below the maximum because of onset of intense absorption of radiation in methane. CH_3^+ ions are quickly converted to $C_2H_5^+$ ions (e17). The lower altitude peak in the electron density profile corresponds to this CH_3^+ ion production. The major ion here is $C_2H_5^+$. We emphasize, however, that CH₃⁺ ion production is purely conjectural and no experimental data or conclusive theoretical estimates of its ionization cross-section and quantum yield are yet available. Maximum p[CH₃⁺] are 1.2 cm⁻³ sec^{-1} and $13 cm^{-3} sec^{-1}$ for $K = 10^5$ and $K \propto M^{-1/2}$ models, respectively; corres-



FIG. 2. Jupiter's model ionosphere with $K = 10^5 \text{ cm}^2 \text{sec}^{-1}$ and $K \propto M^{-1/2}$. CH₃⁺ ion production rates, $p[\text{CH}_3^+]$, for the two models are given by the upper scale. Height scale is the same as in Fig. 1.

ponding Ne at these altitudes are 10^3 cm^{-3} and $2.8 \times 10^3 \text{ cm}^{-3}$. The secondary ion peak due to CH₃⁺ is also dependent upon the value of dissociative recombination coefficient, r_7 ; a value of r_7 twice as large as the one given by Rebbert *et al.* (1973) will essentially wipe out this feature. Improved measurements of r_6 and r_7 are urgently needed.

Discrepancies between the Pioneer 10 results and a pure photochemical ionospheric model involving H, H₂, He, and hydrocarbons but neglecting the possible role of metallic ions and wind shear sporadic *E*-like layering exist and were foreseen as possibilities in the paper by Atreya *et al.* (1974). The mechanisms producing large ion peaks below the principal *F* region peak are being investigated.

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